

1/30/99

II CLAY-WATER FORCES

+Shekhaib

(1.361 Refer.)

1. WATER VAPOR SORPTION ("Adsorbed" Water)

(II-2.1)

1.1 Water Content vs. Relative Humidity (RH)

1) Relative humidity, RH(%)

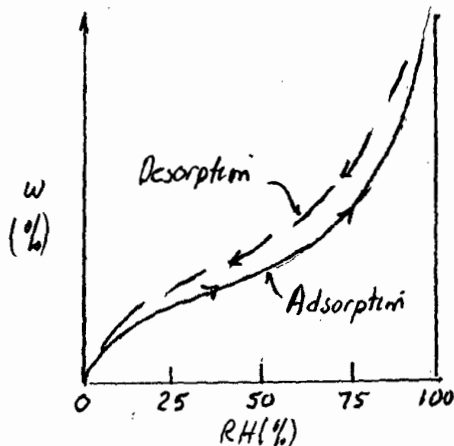
$$RH(\%) = 100 \times \frac{P_w}{P_s} = \frac{\text{partial pressure of (pure) water vapor}}{\text{saturation pressure of pure H}_2\text{O (flat surface) at same temp.}}$$

$$[\log P_s(\text{kPa}) \approx -0.174 + 0.0265 T(^{\circ}\text{C}); T = 20 \pm 15^{\circ}\text{C}]$$

| | | | |
|--------|-----------------------|------|-----------|
| Actual | T = 10 | 20 | 30°C |
| | P _s = 1.23 | 2.34 | 4.245 kPa |

Table 2.7 Fredlund & Rahardjo (1993)
Soil Mech. for Unsaturated Soils, Wiley textbook

2) Adsorption - desorption curves (starting from oven dry soil)



3) Values of water content

a) $RH = 50\% \rightarrow 1-2$ molecular thickness of water : $t = 5\text{\AA}$ $w_{50}(\%) = 0.05 \times SSA(\text{m}^2/\text{g})$

$\phi_{H_2O} = 3\text{\AA}$

$$\{ t(\text{\AA}) = 100 w(\%) / SSA(\text{m}^2/\text{g}) \}$$

• Empirical data from Ladd & Lambe (1961)

$$w_{50}(\%) = (0.2 \pm 0.05) I_p(\%)$$

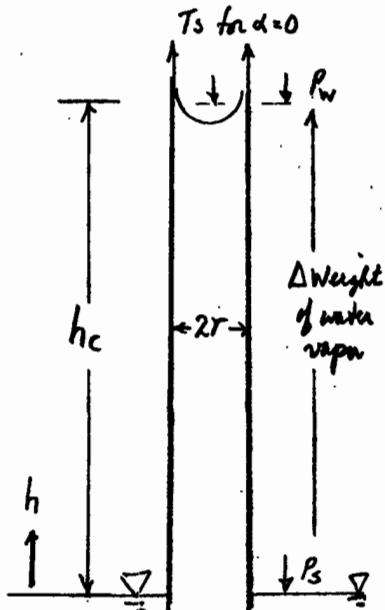
b) $RH = 99\% \rightarrow t = 10-15\text{\AA} \approx 3-5$ molecular thicknesses

Often referred to as thickness of tightly bound "adsorbed" water

(1.361 Refs)
(II-2.1; II-2)

1.2 Capillary Pressure (u_c) vs. Relative Humidity

1) Theoretical relationship



1.361 Eqn. $u_c = h_c \cdot \gamma_w = \frac{2T_s}{r}$, $T_s = \text{surface tension of H}_2\text{O}$
 $= 72.75 \text{ N}\cdot\text{m}/\text{m}^2$
 at 20°C
 (Energy/unit area)

$u_c = u_a - u_w$

Revised eqn. to include reduced vapor pressure of curved meniscus & wgt water vapor

$pV = nR_gT$
 $\frac{dp}{dh} = -\rho_w \gamma$
 mass density of water vapor

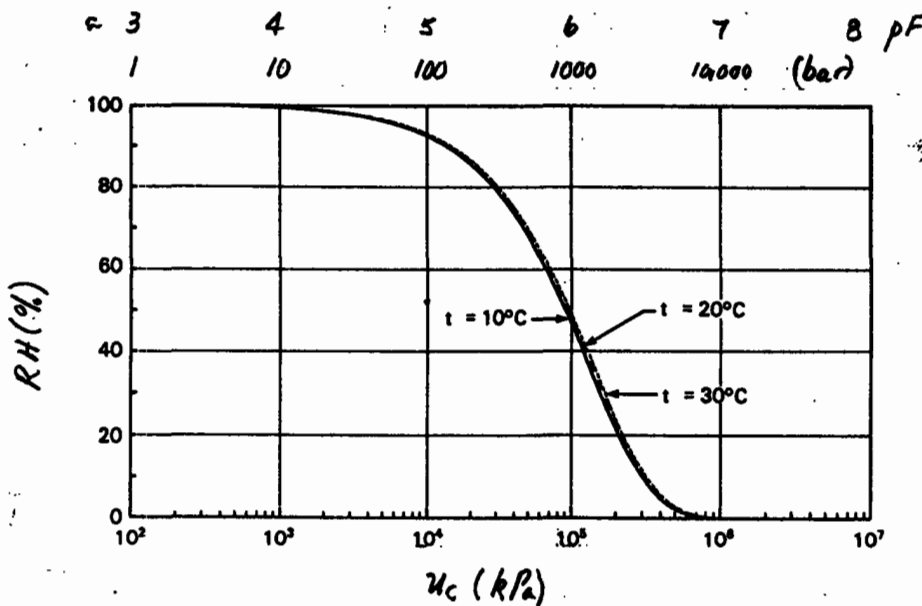
$h_c \cdot \gamma_w = \frac{2T_s}{r} + P_s - P_w$ } At bottom of tube
 Force \downarrow Force \uparrow

$u_c = h_c \cdot \gamma_w = -\frac{\rho_w R_g T}{M} \ln\left(\frac{RH}{100}\right)$

$\rho_w = 998 \text{ kg}/\text{m}^3$
 $R_g = 8.314 \text{ J}/\text{mol}\cdot\text{K}$
 $T = 293 \text{ K}$
 $M = 18.015 \text{ g}/\text{mol}$
 (J = N.m)

For $T=20^\circ\text{C}$ $u_c \text{ (kPa)} = -1.35 \times 10^5 \ln\left(\frac{RH}{100}\right)$

$u_c \text{ (bar)} = -1350 \ln\left(\frac{RH}{100}\right)$



$pF = \log(h_c \cdot \text{cm})$

$h_c = 1000 \text{ cm} = 10 \text{ m}$
 $\rightarrow pF = 3.00 \approx 1 \text{ atm}$
 $\approx 1 \text{ bar} = 100 \text{ kPa}$

$1 \text{ bar} = 100 \text{ kPa} \div$
 $\gamma_w = 9.81 \text{ kN}/\text{m}^3 = 10.2 \text{ m}$
 $= 1020 \text{ cm}$
 $1 \text{ atm} = 1.013 \text{ bar}$

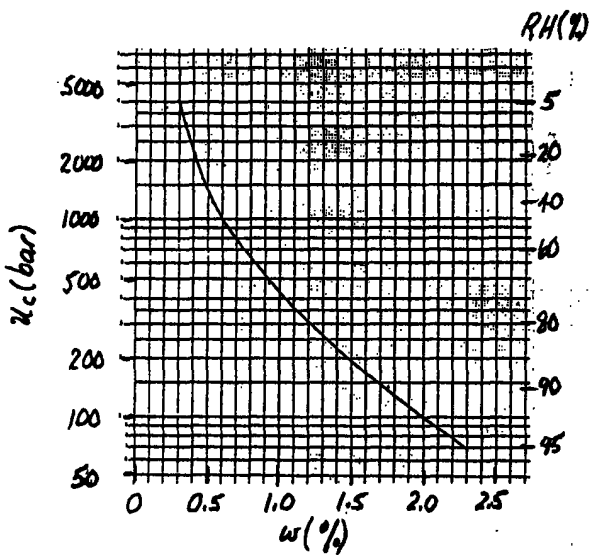
| | | | | | |
|-----------------------|------|------|-------|----|------|
| RH(%) = | 99.9 | 99.5 | 99 | 95 | 90 |
| $u_c \text{ (bar)} =$ | 1.35 | 6.77 | 13.55 | 69 | 142 |
| $pF =$ | 3.15 | | 4.15 | | 5.15 |



(1.2 Cont)

2) Water content vs Capillary pressure

(1.361 Ref)



Computed from w vs RH data on Peerless (Li) Kaolinite with

CEC = 2.7 meq./100g (SSA = 9 m²/g for $\sigma_0 = 0.3$)

Adsorption curve for $RH = 0$ to 95%

w vs RH data by R.F. Martin, 1958, Proc. 5th Nat. Conf. Clays & Clay Mineralogy NAS/NRC Publ. 566, p 23-38

No. 5505 Engineer's Computation Pad



1.3 Mechanisms of Water Vapor Adsorption (CCL + Mitchell (1993), Chap 6)

Note: H₂O molecule = dipole



Starting from overlying soil

Mechanisms

Remarks

(II-2, p6)

1) H-bonding

Very important for 1st layer

2) Cation hydration



Very important: Hydrated dia. of cations $\approx 15 \pm 5 \text{ \AA}$

3) Orientation of H₂O dipoles in electric field

Uncertain importance



(less mobile \rightarrow less free energy)

4) van der Waals attraction

" "

| | | | | | |
|------------------|----------------|-----------------|------------------|------------------|-----------------|
| Cation | K ⁺ | Na ⁺ | Ca ⁺² | Mg ⁺² | Li ⁺ |
| Hydrated Dia (Å) | 9 ± 2.5 | 13.5 ± 2 | 19 | 21.5 | 17.5 ± 2.5 |

(JKM 1993, p122)

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(1.361 Ref)

1.4 Measurement of Water Content

- 1) ASTM 2215 $T = 110 \pm 5^\circ\text{C}$ really doesn't remove all H_2O ; need maybe 200°C
Is that of practical significance? _____
- 2) Don't leave soil in humid environment before weighing after remove from oven

1.5 Tensile Strength of Water [Ridley & Burland 1993, geot 43(2)]

(Can $u_c \rightarrow 10$'s to 100 's atm. in cohesive soils? _____)

- 1) Tabor (1979): theoretical tensile strength, $u_w = -5000 \text{ atm}$.
However, very small amount of dissolved gas \rightarrow cavitation at
 $u_w \gg$ theoretical value ($u_c \ll 5000 \text{ atm}$). Typical lab test \rightarrow ? _____

2) Temperley & Chambers (1946 Proc. Royal Soc., London)



Carefully degassed, smooth-walled chamber $\rightarrow u_w = -5000 \text{ atm}$

- 3) Conclusion: answer to () is YES, e.g. drying of initially sat. clay to the shrinkage limit
(see p8)

2. SOIL SUCTION (S) [Also see 1.361 Section 2.6 of Part II-1]

2.1 Overview

- Used as a measure of the "free energy" of the bulk pore water in soil* (at constant elevation) relative to pure water at atm. pressure at same temperature when this free energy is less than zero.

- Free energy can be expressed in terms of:

| | | | |
|-----------|--------|---|-------------|
| potential | - J/kg | } | most common |
| pressure | - kPa | | |
| head | - m | | |

- \rightarrow * Questions: (1) Restricted to soil with $\sigma - u_a = 0$?
(2) Physical meaning when low % sat. \rightarrow no bulk water?

22-141 50 SHEETS
22-142 100 SHEETS
22-144 200 SHEETS



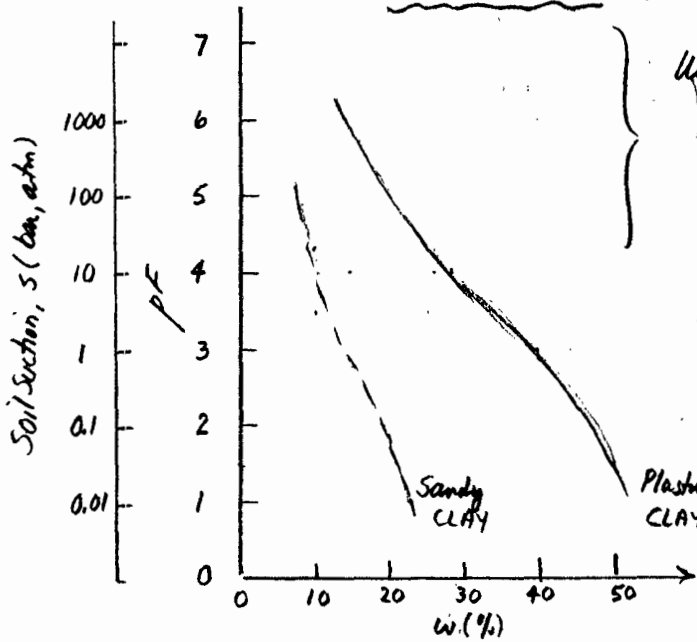
CCL tentative definition

$S =$ free energy of pure H_2O in contact with soil having $\sigma - u_a = 0$

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2.1 Cont

Soil suction curves (location = f wetting vs drying)



Usually indirect, e.g. relative humidity (psychrometer) or filter paper
(CCL believes that most data are not very reliable)

Various direct measurements, e.g. suction plate ($u_a = 0$), pressure plate ($u_c = u_a - u_w$), tensiometer = piezometer with infinite porous stone
à la Section 2.4

Note: Most reported data are for drying; will show wetting/drying cycles when cover compacted soils. Also usually use $\theta_w =$ volumetric water content = V_w/V (p8)

2.2 Components of Soil Suction (Aitchison & Richards, 1965) (1361 II2, p8)

- 1) Total suction (s) = matric suction (s_m) + solute suction (s_s)
- 2) Solute suction (s_s) = osmotic pressure due to salt concentration in bulk pore water of soil
 s_s (atm) ≈ 24 (salt conc., M = moles/l) for both cations & anions for $T = 20^\circ C$
(Sea water ≈ 35 g/l $\rightarrow \approx 1.1$ M $\rightarrow \approx 26$ atm)

matrix = matrix

- 3) Matric suction (s_m) = $(u_a - u_w)$, where pore water in measurement system has same salt conc. as bulk pore water in soil

NOTE: In CCL opinion, matric suction denotes s_m for soil having $\sigma - u_a = 0$ (at least via techniques used in lab to measure s_m). We should discuss difference between s_m & σ'

2.3 Mechanisms Causing Matric Suction

- 1) Those for water vapor sorption in Section 1.3 (i.e. for adsorbed H_2O)
- 2) Osmotic pressure due to depressed double layer, i.e., need apply neg. u_w to prevent soil from sucking in more water (≠ solute suction)
- 3) Elastic deformation of platy particles (mechanical) drying \rightarrow bonding, wetting \rightarrow unbonding
- 4) Net contact stress ($\bar{\sigma}'_{ac}) > 0$

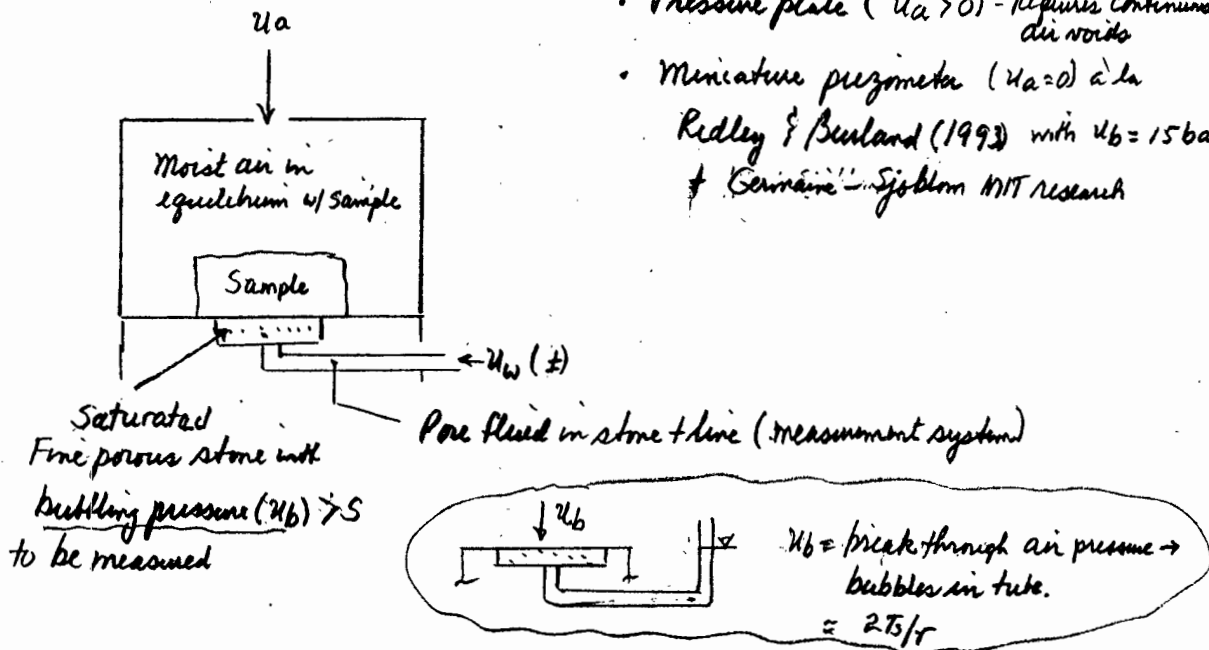
need $2d > 10 - 15 \text{ \AA}$



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2.4 Direct Lab Measurements of Soil Suction (Sheet A)

- 1) Experimental set-up for:
 - Suction plate ($u_a = 0$) "axis translation"
 - Pressure plate ($u_a > 0$) - requires continuum of voids
 - Miniature piezometer ($u_a = 0$) a la Ridley & Burland (1993) with $u_b = 15 \text{ bar}$ & Germaine - Sjöblom MIT research



2) Test conditions & measured soil suction = ($u_a - u_w$)

| Sample | Case | System Pore Fluid | Does Stone Act as Semi-Perm. Membrane? | Meas. Soil Suction | Remarks |
|-----------------------------------|------|-------------------------|---|--------------------|---------|
| Clay with salt in bulk pore fluid | 1 | Same as soil pore fluid | Both Yes & No, i.e. makes no difference | Matric | — |
| " | 2a | Pure H_2O | Yes | Total | } ** |
| " | 2b | " | No | Matric | |

** Whether or not 15 bar stone will exhibit a significant "osmotic efficiency" (i.e., act as partial semi-permeable membrane) is somewhat controversial. For example, discussion⁽¹⁾ to Ridley & Burland concludes that 15-bar stone approaches total suction, whereas authors disagree. Kurt Sjöblom's^(9/00) data indicate that measure matric suction

(1) Geotechnique 1994, 44(3), 551-556

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2.5 Soil Suction Measurement Techniques

- 1) See Sheet A for overview & schematic of direct measurements
- 2) " " B for info. on filler paper technique *
- 3) See Stannard [1992: ASTM, GTJ, 15(1)] for 50A paper on field measurement technique

* Note that calibration curve becomes very flat (hence very imprecise) at low suctions. Also want details of "contact" vs "noncontact" very important, plus may have different calibration curves for these 2 conditions (RSB, 1994 closure)

3. NATURE OF ADSORBED WATER

3.1 Total vs. Pressure Head and Attraction Pressure

- 1) Components of total head (total free energy)

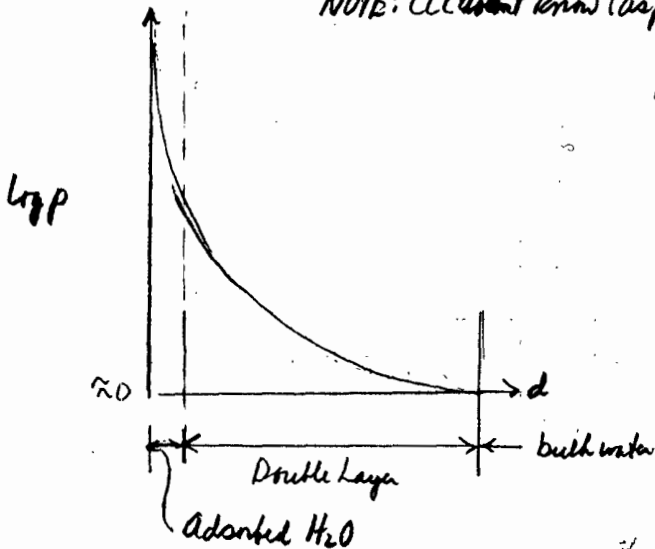
$$\text{Total head } h = h_t = h_e + h_p + h_s + h_v + h_{pc}$$

$\begin{matrix} \frac{s_s}{\rho_w} & \frac{1}{2} v^2 \\ \text{at } h_e = h_s = h_v = 0 \end{matrix}$

physics-chemical within double layer (DL) (JKM 1993 Section 9.5)

- 2) Attraction pressure^(p), vs. distance^(d), from particle surface

NOTE: CCL don't know (as per writing) shape of curve



• Discussion of actual pressure in H₂O → can have very high s_m, but still have large h_p in adsorbed water

• Values of p within adsorbed H₂O

$d = 10-15 \text{ \AA} \rightarrow 10^5 \text{ atm}$

$d = 3-6 \text{ \AA} \rightarrow 10^3-10^6 \text{ atm}$

- 3) Discussion: what is meaning of h_t, h_p, s_m etc. when low % sat. → no bulk water?

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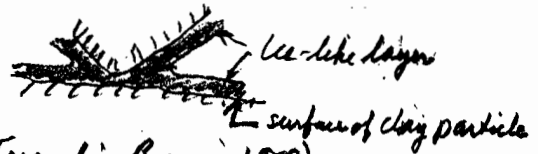
(1.361 Ref)

(II, p6)

3.2 Physical Properties of Adsorbed Water

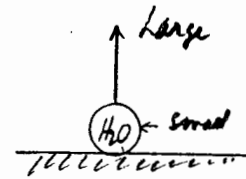
1) Is "ice-like"

- Very high viscosity
- Causes cohesion & creep (e.g., Terzaghi, Bjerrum 1973)
- Don't get mineral-mineral contacts (i.e. greatly inhibits) in clay



2) Like 2-D liquid (R.T. Martin)

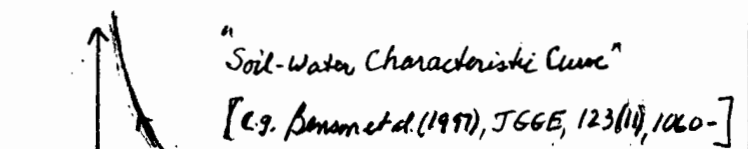
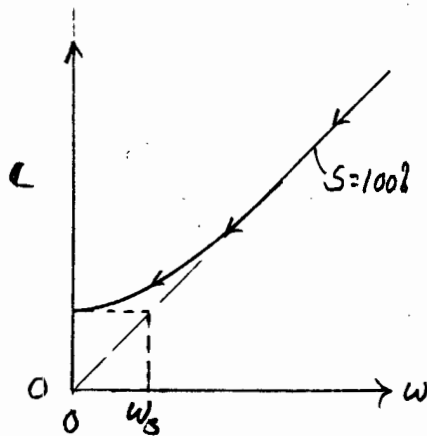
- Like ball bearings on magnet
- Does not cause strength, but does inhibit min.-min. contact



3) Discussion -

- Story of discussion at 1961 ICSMFE, Paris (Gueze of Holland: "adsorbed H₂O → clay strength → creep." TWL reply =

- Predicted unconfined s_w of initially saturated clay during drying



$$\begin{aligned} \text{Volumetric w.c.} = \theta_w &= \frac{V_w}{V} = \frac{S V_v}{V} = S \cdot n \\ &= \frac{S e}{1+e} = \frac{G_s w}{1+G_s w/S} \end{aligned}$$

• From 1.361

- Strength of oven-dried BBC & what happened when put in water
- Part II-1, p5 → $\sigma'_{int} = 70 \text{ bar}$ at shrinkage limit of BBC

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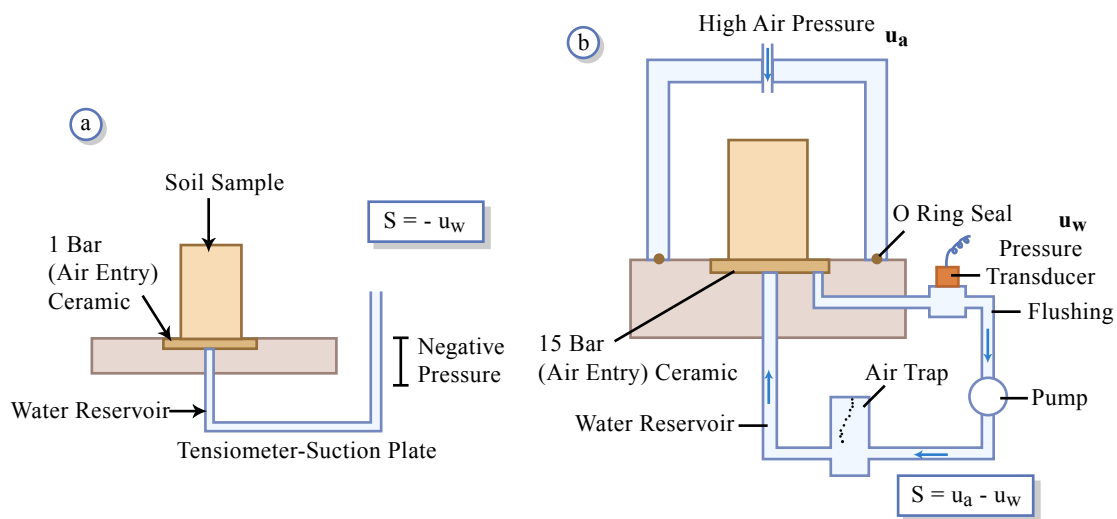
1.322 Part A-D

Adapted from RIDLEY AND BURLAND (1993)

Suction Measurement Techniques

| | Suction Value [#] | Principal Usage | Direct/ Indirect | Range: kPa | Equilibrium Time |
|---------------------|----------------------------|-----------------|------------------|----------------------------------|------------------|
| Vacuum Desiccator | Total | Lab. | Indirect | 10 ³ -10 ⁶ | Months |
| Psychrometer | Total | Field | Indirect | 300-7000 | Months |
| Filter Paper | Total | Field | Indirect | 1000-30000 | Weeks |
| | Matrix | Lab. | Indirect | 30-30000 | 1 week |
| Porous Block | Matrix | Field | Indirect | 30-3000 | Weeks |
| Thermal Block | Matrix | Field | Indirect | 0-175 | Days |
| Suction Plate | Matrix | Lab. | Direct | 0-90 | Hours |
| Tensiometer | Matrix | Field | Direct | 0-90 | Hours |
| Pressure Plate | Matrix | Lab. | Direct | 0-5000 | Hours |
| Osmotic Tensiometer | Matrix | Field | Direct | 0-1500 | Days |

As defined by Aitchison and Richards (1965).



Direct Measurement of Soil Suction: a) Tensiometer; b) Pressure Plate Apparatus

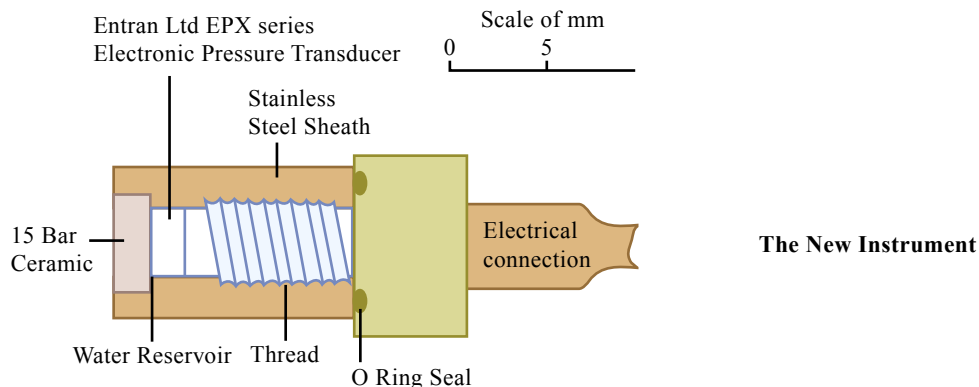


Figure by MIT OCW.

Adapted from:

Geotechnique (1993) 43(2), 321-324

Also see discussion by Mesriho & Chandler (1999) & discussion by Ridley & Burland - Geot 44(3), 551-556

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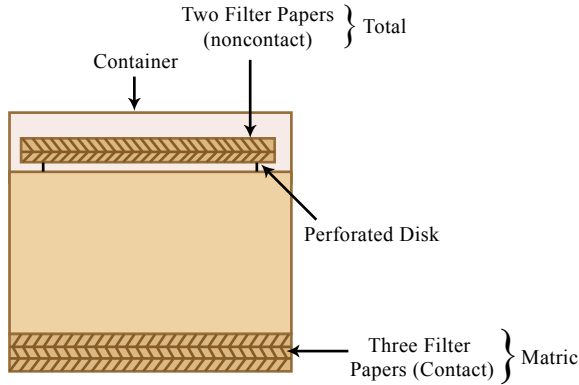
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1.322 Part A-II

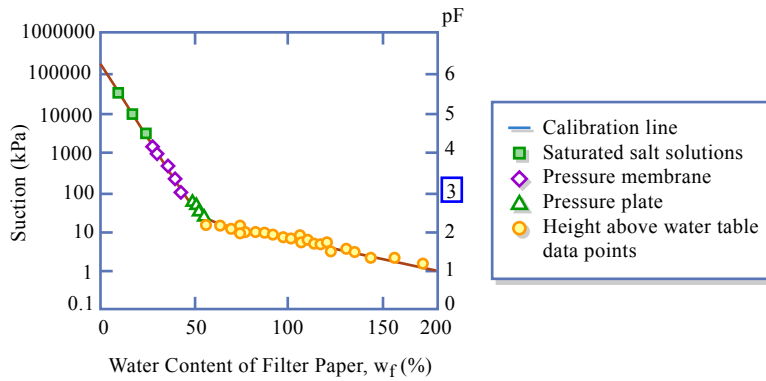
Adapted from Fredlund & Rahardjo (1993) *Soil Mechanics for Unsaturated Soils*,
John Wiley & Sons

John Wiley & Sons

Filter paper technique



Contact and noncontact filter paper methods for measuring matric and total suction, respectively.



A typical calibration curve showing measured filter paper water contents for applied suctions.

Figure by MIT OCW.

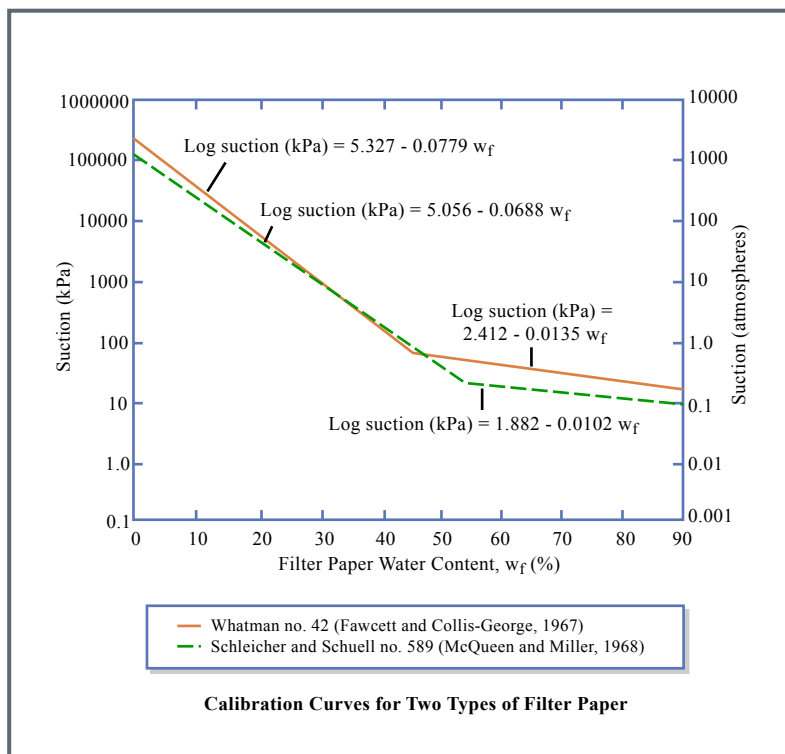


Figure by MIT OCW.

B

50 SHEETS
100 SHEETS
200 SHEETS

